

Notes:

- a. For Unknown Samples, we recommend running several dilutions of the samples to ensure that the readings are within the Standard Curve range.
 - b. Do not re-use the diluted PC Positive Control.
- 2. Standard Curve Preparation:** Add 0, 2, 4, 6, 8, 10 μl of 10 mM NADH Standard stock solution into a series of wells to generate 0, 20, 40, 60, 80, 100 nmole/well of NADH Standard respectively. Adjust the volume to 100 μl /well with PC Assay Buffer.
- Note:** Standards can be prepared at the end of the incubation time.
- 3. Substrate Hydrolysis:** Mix enough reagents for the number of assays to be performed. Prepare 50 μl of Reaction Mix as indicated in the table below:

	<u>Reaction Mix</u>	<u>Background Mix</u>
PC Assay Buffer	34 μl	36 μl
NADH Standard	8 μl	8 μl
PC Developer	2 μl	2 μl
PC Substrate II	2 μl	2 μl
PC Cofactor	2 μl	2 μl
PC Substrate I	2 μl	--

Add 50 μl of Reaction Mix to each well containing Sample, Positive Control, and Reagent Background Control and 50 μl of Background Mix to Sample Background Control wells and mix well, protected from light. The total volume of each well including Sample, Sample Background Control, Positive Control, and Reagent Background Control is 100 μl .

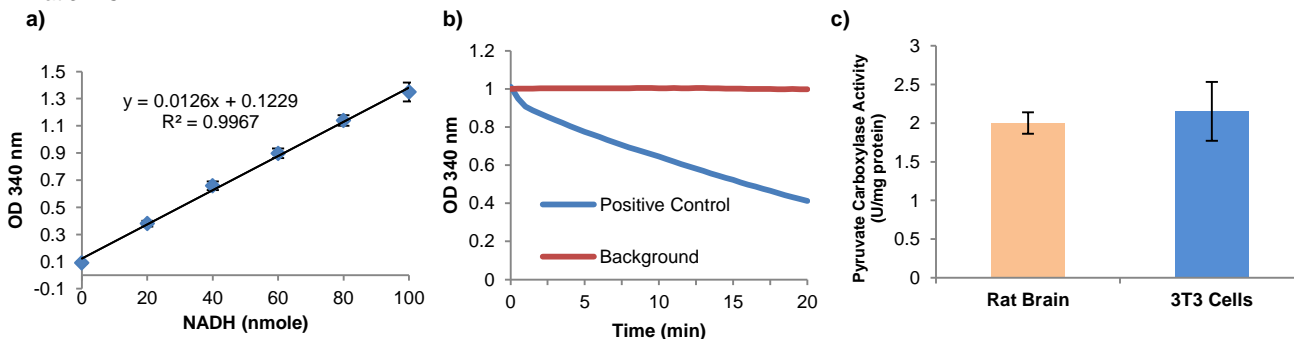
Note: Prepare Reaction Mix and Background Mix immediately before adding to the wells.

- 4. Measurement:** Measure the absorbance of all wells at 37 °C for 20 min in kinetic mode at 340 nm. NADH Standard can be read in endpoint mode.
- 5. Calculation:** Subtract 0 Standard reading from all Standard readings. Plot the NADH Standard Curve. Choose any two time points (t_1 & t_2) within the linear portion of the curve for each Sample type and obtain the corresponding absorbance values (OD_1 and OD_2 ; $\text{OD}_1 > \text{OD}_2$). Subtract OD_1 and OD_2 from the OD values of the corresponding Sample Background control well for the same time points. Apply the corrected Sample readings to the NADH Standard Curve to get B nmole of NADH consumed during the reaction time ($\Delta t = t_2 - t_1$).

$$\text{Sample Pyruvate Carboxylase Activity} = B \times D / (\Delta t \times V \times M) = 1000 \times \text{nmole} / \text{min} / \text{mg} = U / \text{mg}$$

Where: **B** = Amount of NADH consumed, calculated from the NADH Standard Curve (in nmole)
 $\Delta t = t_2 - t_1$ (in min)
V = Sample volume added into the reaction well (μl)
M = Initial Sample concentration (mg protein / ml)
D = Sample dilution factor ($D = 1$ for undiluted samples)

Unit Definition: One unit of Pyruvate Carboxylase activity is the amount of enzyme that consumes 1.0 μmole of NADH per min, at pH 7.4 at 37 °C.



Figures: (a). NADH Standard Curve, results from multiple experiments. (b). Pyruvate Carboxylase activity of Positive Control. (c). PC activity in Rat Brain (2 μg protein) and 3T3 cells (2 μg protein) lysates. All assays were performed following kit protocol.

VIII. Related Products:

- Pyruvate Carboxylase, Rhizobium etli Recombinant (P1661)
- NADH Disodium salt trihydrate (2735)
- Pyruvate Colorimetric/Fluorometric Assay Kit (K609)
- Malate Dehydrogenase Activity Colorimetric Assay Kit (K654)
- Dounce Tissue Homogenizer (1998)

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